# GLOBAL SENSITIVITY ANALYSIS OF NONLINEAR CHEMICAL KINETIC EQUATIONS USING LIE GROUPS: I. DETERMINATION OF ONE-PARAMETER GROUPS 

C.E. WULFMAN<br>Department of Physics, The University of the Pacific, Stockton, CA 95207, USA and

H. RABITZ

Department of Chemistry, Princeton University, Princeton, NJ 08544, USA
Received 14 December 1987
(in final form 2 January 1989)


#### Abstract

We introduce one-parameter groups of transformations that effect wide-ranging changes in the rate constants and input/output fluxes of homogeneous chemical reactions involving an arbitrary number of species in reactions of zero, first and second order. Each one-parameter group is required to convert every solution of such elementary rate equations into corresponding solutions of a one-parameter family of altered elementary rate equations. The generators of all allowed oneparameter groups are obtained for systems with $N$ species using an algorithm which exactly determines their action on the rate constants, and either exactly determines or systematically approximates their action on the concentrations. Compounding the one-parameter groups yields all many-parameter groups of smooth time-independent transformations that interconvert elementary rate equations and their solutions.


## 1. Introduction

The response of kinetic systems over extensive regions of their physical parameter space - the space of rate constants and input/output fluxes - is of wide interest in many different contexts. For example, chemical system modelling can involve solving large numbers of coupled rate equations with considerable uncertainties in many values of the rate constants. In other problems some of the system parameters (e.g. input fluxes of chemical species) may actually be controlled, but determing the optimum choice of parameter values would require exploring a large domain of
control-parameter space. Conventional gradient-based local sensitivity analysis techniques [1] have limited applicability in problems of this type. In addition, fully statistically-based approaches [2] do not allow for an analysis of the structure of the parameter space. Other methodologies [3] based on repeated sampling of points in the parameter space suffer from the same problem and often require an impractical amount of computational labor.

In two previous papers, an alternative approach to sensitivity analysis, using Lie transformation groups, was introduced as a method for investigating the consequaneces of large changes in parameters in kinetic equations [4,5]. The present paper extends this effort into the realm of nonlinear kinetics.

The thrust of this work is the development of a systematic procedure that yields mappings which transform solutions of a system of kinetic equations through the hyperdimensional space defined by all rate constants, chemical species, and time. Here we will not, however, consider transformations of the time variable. We also do not allow the transformed rate constants to be explicit functions of the concentration variables.

The mappings are achieved by the application of operators $T(a)=\exp (a U)$ of one-parameter groups, where $a$ is a real parameter and $U$ is a group generator of Lie type. This generator is a first-order differential operator which may act on all physical parameters and variables of the kinetic system. Symbolizing concentrations by $x_{i}$ and rate constants by $k_{\mu}$, the generator here takes the form

$$
\begin{equation*}
U=\sum h_{i}(x, k) \partial / \partial x_{i}+\sum g_{\mu}(k) \partial / \partial k_{\mu} \tag{1.1}
\end{equation*}
$$

Here, $x$ represents the set of $x_{i}$ and $k$ represents the set of $k_{\mu}$. Henceforth, $\boldsymbol{x}, \boldsymbol{k}$ represent vectors with components $x_{i}$ and $k_{\mu}$ in a Euclidean space of $x, k$. The operator of finite transformations $T(a)=\exp (a U)$ acts as follows:

On a rate constant $k_{\mu}$ :

$$
\begin{equation*}
T(a) k_{\mu}=\bar{k}_{\mu}=K_{\mu}(k ; a) ; K_{\mu}(k ; 0)=k_{\mu} \tag{1.2a}
\end{equation*}
$$

On the concentration $x_{i}$ of species $i$ :

$$
\begin{equation*}
T(a) x_{i}=\bar{x}_{i}=X_{i}(\boldsymbol{x}, \boldsymbol{k} ; a) ; X_{i}(\boldsymbol{x}, \boldsymbol{k} ; 0)=x_{i} \tag{1.2b}
\end{equation*}
$$

Figure 1.1 depicts the type of mapping being considered.
As indicated in eqs. (1.2), assigning the group parameter $a$ the value zero gives the identity transformation. As $a$ is shifted from zero by infinitesimal and then finite amounts, changes in $k$ and $\boldsymbol{x}$ develop which are at first infinitesimal, and then become increasingly profound. For a fixed value of the parameter $a, T(a)$ acts on the moving vector $x(t)$ to give the transformed vector $\bar{x}(t)=X(x(t), k ; a)$. It thus transforms the curve in concentration space described by $\boldsymbol{x}(t)$ into a new curve depict-


Fig. 1.1. The mappings in $x, k, t$ space. The mappings $P \rightarrow \bar{P}$ represent the concentration changes $\boldsymbol{x} \rightarrow \bar{x}$ and the changes in rate constants $k \rightarrow \bar{k}$, while the time $t$ is held fixed. As $\bar{k}$ is not a function of $x$ or $t$, the trajectory $P \rightarrow P^{\prime}$ is mapped into a trajectory $\bar{P} \rightarrow \bar{P}^{\prime}$ that lies in a hyperplane of constant $\bar{k}$.
ing an altered evolution of chemical concentrations. By changing the value of the parameter $a$, one is able to convert an initial evolution curve into a one-parameter family of evolution curves. Thus, in fig. 1.1 the upper curve may be considered as one member of a family of transformed curves, a curve obtained by giving the group parameter a specific value. The value of the group parameter $a$ can be assigned by the investigator, but it is neither a rate constant nor a concentration. Its chemical significance is determined by the functions $K_{i}$ and $X_{\mu}$ in (1.2). This significance, and that of the generator $U$, can be assessed by investigating the action of the operator of the infinitesimal transformation $T(\delta a)$.

Letting $a \rightarrow \delta a$, one has

$$
\begin{equation*}
\exp (a U) \rightarrow \exp (\delta a U) \sim 1+\delta a U \tag{1.3}
\end{equation*}
$$

Thus, for an infinitesimal transformation,

$$
\begin{equation*}
\bar{x}_{i}=x_{i}+\delta a U x_{i}=x_{i}+\delta a h_{i}(x, k) ; \bar{k}_{\mu}=k_{\mu}+\delta a U k_{\mu}=k_{\mu}+\delta a g_{\mu}(k) \tag{1.4}
\end{equation*}
$$

Consequently, if one defines $\delta x_{i}$ as $\bar{x}_{i}-x_{i}$ and $\delta k_{\mu}$ as $\bar{k}_{\mu}-k_{\mu}$ in (1.4) one has

$$
\begin{equation*}
\delta x_{i}=\delta a h_{i}(x, k), \quad \delta k_{\mu}=\delta a g_{\mu}(k) . \tag{1.5}
\end{equation*}
$$

It follows that $T(\delta a)$ changes the concentration $x_{i}$ by an amount $\delta a h_{i}$ that may depend upon all concentrations $x$ and rate constants $k$. Similarly, the transformation changes the rate constant $k_{\mu}$ by an amount $\delta a g_{\mu}$ that may depend upon all rate constants $k$. As an example, consider the generator

$$
\begin{equation*}
U=k_{11} x_{1} \partial / \partial x_{1}+2 \partial / \partial k_{01} \tag{1.6}
\end{equation*}
$$

and its action on a system involving a single species obeying the rate equation

$$
\begin{equation*}
\mathrm{d} x_{1} / \mathrm{d} t=k_{10}+k_{11} x_{1}+k_{111} x_{1}^{2} . \tag{1.7}
\end{equation*}
$$

This generator determines a shift in the concentration $x_{1}$ by an amount $\delta x_{1}=k_{11} x_{1} \delta a$, i.e. a shift proportional to the product of the concentration and the second-order rate constant. This determines a consequent shift in $\mathrm{d} x_{1} / \mathrm{d} t$ by an amount $\mathrm{d}\left(k_{11} x_{1} \delta a\right) / \mathrm{d} t$ $=\delta a k_{11} \mathrm{~d} x_{1} / \mathrm{d} t$. It also determines a shift $\delta k_{10}=2 \delta a$ in the flux $k_{10}$. The generator does not affect either $k_{11}$ or $k_{111}$.

Now, if it were true that the shifted concentration obeyed the same rate equation with the shifted value of $k_{10}$, the generator (1.6) could be of use in investigations of the consequences of changing the rate of supply or removal of the reagent. The operator $T(a)=\exp (a U)$ could then be used to determine the relation between changes in the flux and changes in the concentration $x$, the extent of both changes being determined by the value of the parameter $a$. However, the $U$ of (1.6) was chosen at random and can not be expected at each value of $t$ to convert $x(t)$ into $\bar{x}(t)$ that obey the altered rate equation.

If the $U$ of (1.6) had the property that $U F=0$, where

$$
\begin{equation*}
F=\left(k_{10}+k_{11} x_{1}+k_{111} x_{1}^{2}\right), \tag{1.8}
\end{equation*}
$$

then $\exp (a U)$ acting on the right-hand side of (1.7) would leave it unchanged, i.e. not change the reaction rate. This is because

$$
\begin{equation*}
\exp (a U) F=\left(1+a U+\frac{1}{2} a U a U+++\right) F \tag{1.9a}
\end{equation*}
$$

would then give

$$
\begin{equation*}
F+0+0+++=F . \tag{1.9b}
\end{equation*}
$$

This is not, however, the restriction we wish to impose.

The restrictions we impose upon the $T(a)$, and hence the $U$ 's, so as to obtain chemical information from them are as follows: Each $T(a)$ will be required to have a unique action on all $k$, $x$, in an elementary kinetic equation, map contiguous values of $k_{\mu}$ and $x_{i}$ into contiguous values of $\bar{k}_{\mu}$ and $\bar{x}_{i}$, and give $\bar{k}$ and $\bar{x}$ that also satisfy elementary kinetic equations (cf. section 2 below). In addition, we shall require that all the variables $a, x, k$ are real. Taken together, these requirements ensure that the transformation $T(a)$ maps solutions of the set of kinetic equations

$$
\begin{equation*}
\mathrm{d} x_{i} / \mathrm{d} t=k_{i 0}+k_{i j} x_{j}+k_{i j j^{\prime}} x_{j} x_{j^{\prime}} \tag{1.10a}
\end{equation*}
$$

into solutions of the set of transformed equations

$$
\begin{equation*}
\mathrm{d} \bar{x}_{i} / \mathrm{d} t=\bar{k}_{i 0}+\bar{k}_{i j} \bar{x}_{j}+\bar{k}_{i j j^{\prime}} \bar{x}_{j} \bar{x}_{j^{\prime}} \tag{1.10b}
\end{equation*}
$$

They impose restrictions on the form of the generators $U$ sufficient to ensure that the $U$ may be determined algorithmically. Because of this, one has available a systematic method for investigating the manner in which changes in rate constants are associated with changes in species concentrations and their time evolution. These restrictions are not equivalent to requiring that $T(a)$ leave reaction rates $\mathrm{d} x_{i} / \mathrm{d} t$ invariant.

In the next section, we outline an algorithm for determining the allowed Lie generators $U$ and use it to completely determine the terms in the generators which govern the transformation of rate constants of kinetic systems with an arbitrary number of species. The remaining terms in the generators, governing the transformation of species concentrations, are approximated by power series whose zero-, first-, and second-order terms we determine.

## 2. Derivation of approximate invariance operators: Their action

Let a set of kinetic equations be given as

$$
\dot{x}=r(x, k)
$$

with

$$
\begin{align*}
\dot{x}= & \mathrm{d} x / \mathrm{d} t ;-\infty<t, x_{i}, \dot{x}_{i}<\infty \\
r= & \left(r_{1}, r_{2}, \ldots\right) \\
r_{i}= & k_{i 0}+k_{i j} x_{j}+k_{i j j^{\prime}} x_{j} x_{j^{\prime}}, \quad j^{\prime} \geqslant j \\
& i, j, j^{\prime}=1,2, \ldots ; \quad-\infty<k_{\mu}<\infty \tag{2.1}
\end{align*}
$$

The evolution operator of this system is then $\exp (t V)$, with

$$
\begin{equation*}
V=r \cdot \nabla_{x}, \quad \nabla_{x}=\left(\partial / \partial x_{1}, \partial / \partial x_{2}, \ldots\right) \tag{2.2a}
\end{equation*}
$$

That is,

$$
\begin{equation*}
\vec{x}=\exp (t V) x=X(x, k ; t) \tag{2.2b}
\end{equation*}
$$

is the vector that $x$ evolves into after a time interval $t$.
Define the operator $\exp (a U)$ of a one-parameter Lie group of transformations with real parameter $a,(-\infty<a<\infty)$ and generator $U$ of the form

$$
U=h \cdot \nabla_{x}+g \cdot \nabla_{k}
$$

where

$$
\begin{align*}
& h=\left(h_{1}, h_{2},,,\right), h_{i}=h_{i 0}+h_{i j} x_{j}+h_{i j j^{\prime}} x_{j} x_{j^{\prime}}+++ \\
& h_{i j j^{\prime}}=h_{i j^{\prime} j}, \text { etc. }  \tag{2.3}\\
& g \cdot \nabla_{k}=\sum g_{i m} \partial / \partial k_{i m}, \quad m=0, j, i j^{\prime} \ldots .
\end{align*}
$$

Here, and in the remainder of the paper, we use the index $m$ in $h_{i m}, k_{i m}$ and $g_{i m}$ to signify any of the values $0, j, j j^{\prime} \ldots$.

The coefficients $h_{i m}$ may in general be allowed to be explicit functions of $t$, $\boldsymbol{x}, \boldsymbol{k}$. The coefficients $g_{i m}$ are not allowed to depend upon $\boldsymbol{x}$ or $t$ but can depend upon $\boldsymbol{k}$. In ref. [4] it was shown that with these restrictions the action of $\exp (a U)$ on the variables $\boldsymbol{x}$ and $\boldsymbol{k}$ is to give a set of transformed variables $\bar{x}$ and $\bar{k}$ in which the $\overline{\boldsymbol{k}}$ have fixed values that do not change with time, while the $\bar{x}$ are, like the $x$, running variables whose values change with time. On transformation, the new values of the $k_{i m}$ depend upon the old values, but not upon $x$ or $t$ : geometrically, the space of the $k_{i m}$ is an invariant subspace of the space of $\boldsymbol{x}, t, \boldsymbol{k}$. The $k_{i m}$ are allowed to take on any real values, and in particular may take on the special value zero without altering the general form of the equations given in (2.1).

It was also shown in ref. [4] that the transformed equations will be of the same general form, (2.1), with $\boldsymbol{x}$ replaced by $\overline{\boldsymbol{x}}$ and $\boldsymbol{k}$ replaced by $\overline{\boldsymbol{k}}$ if and only if

$$
\begin{equation*}
W \equiv[V, U]+\partial U / \partial t=0 \tag{2.4}
\end{equation*}
$$

In this paper, we shall require that the $h_{i m}$ are time independent so that here $\partial U / \partial t$ is zero. $W$ is then easily seen to be of the form

$$
W=w \cdot \nabla_{x}
$$

with

$$
\boldsymbol{w}=\left(w_{1}, w_{2} \ldots\right)
$$

and

$$
\begin{equation*}
w_{i}=w_{i 0}+w_{i j} x_{j}+w_{i j j^{\prime}} x_{j} x_{j^{\prime}}+++ \tag{2.5}
\end{equation*}
$$

For (2.4) to hold in the time-independent case, it is necessary that each of the coefficients $w_{i m}$ vanish identically. For reasons explained below, we shall at first approximate $h$ by the terms explicitly listed in (2.3) and only require that the coefficients given explicitly in (2.5) vanish. The resulting quadratic approximation to the generators $U$ will later be improved by methods discussed in the succeeding paper II. Each $w_{i m}$ in (2.4) is a bilinear function of the $k_{i m}$ and $h_{i m}$, and is linear in the $g_{i m}$. Our first problem is to determine the $h_{i m}$ and the $g_{i m}$.

Before determining the generators in which $\boldsymbol{h}$ is quadratically approximated, it is helpful to understand the effect of allowing $h$ to depend upon polynomials of arbitrary degree in the $x_{i}$. To this end, we classify the contributions to $U, V, W$ according to their degree in $x$. We write

$$
\begin{equation*}
r=r^{(0)}+r^{(1)}+r^{(2)} \tag{2.6}
\end{equation*}
$$

where $r^{(p)}$ is a homogeneous polynomial of degree $p$ in $x$, and we write

$$
V^{(p-1)}=r^{(p)} \cdot \nabla_{x}
$$

to indicate that the corresponding contribution to the generator is of one degree less. Then

$$
\begin{align*}
V & =\left(r^{(0)}+r^{(1)}+r^{(2)}\right) \cdot \nabla_{x}=V^{(-1)}+V^{(0)}+V^{(1)} \\
U & =\left(h^{(0)}+\boldsymbol{h}^{(1)}+\boldsymbol{h}^{(2)}+\boldsymbol{h}^{(3)}+++\right) \cdot \nabla_{x}+g \cdot \nabla_{k} \\
& =U^{(-1)}+U^{(0)}+U^{(1)}+U^{(2)}+++g \cdot \nabla_{k} \\
W & =[U, V]=W^{(-1)}+W^{(0)}+W^{(1)}+W^{(2)}+++. \tag{2.7}
\end{align*}
$$

Now the commutator of $U^{(m)}$ and $V^{(n)}$ is of degree $m+n$, and the commutator of $k \cdot \nabla_{g}$ and $V^{(n)}$ is of degree $n$. Thus, the vanishing of $W$ requires that

$$
\begin{align*}
& 0=W^{(-1)}=\left[U^{(-1)}, V^{(0)}\right]+\left[U^{(0)}, V^{(-1)}\right]+g \cdot \nabla_{k}\left(r^{(0)} \cdot \nabla_{x}\right)  \tag{2.8a}\\
& 0=W^{(0)}=\left[U^{(-1)}, V^{(1)}\right]+\left[U^{(0)}, V^{(0)}\right]+\left[U^{(1)}, V^{(-1)}\right]+g \cdot \nabla_{k}\left(r^{(1)} \cdot \nabla_{x}\right)  \tag{2.8b}\\
& 0=W^{(1)}=\left[U^{(0)}, V^{(1)}\right]+\left[U^{(1)}, V^{(0)}\right]+\left[U^{(2)}, V^{(-1)}\right]+g \cdot \nabla_{k}\left(r^{(2)} \cdot \nabla_{x}\right)  \tag{2.8c}\\
& 0=W^{(2)}=\left[U^{(1)}, V^{(1)}\right]+\left[U^{(2)}, V^{(0)}\right]+\left[U^{(3)}, V^{(-1)}\right]  \tag{2.8~d}\\
& 0=W^{(p)}=\left[U^{(p-1)}, V^{(1)}\right]+\left[U^{(p)}, V^{(0)}\right]+\left[U^{(p+1)}, V^{(-1)}\right], p>3 . \tag{2.8e}
\end{align*}
$$

Note that each of these equations stands for a set of separate equations $w_{i m}=0$, where $w_{i m}$ is the coefficient of

$$
\begin{equation*}
\partial / \partial x_{i}, x_{j} \partial / \partial x_{i}, x_{j} x_{j^{\prime}} \partial / \partial x_{i}-- \text { as } m=0, j, j j^{\prime} \ldots \tag{2.9}
\end{equation*}
$$

A key feature of the set of equations $w_{i m}=0$ is the fact that their rank is much less than their order, so that their solution contains many free parameters. If we do not allow cubic and higher degree polynomials in $x$ into $U$ and $W$, we find that the equations $w_{i m}=0$ for $W^{(-1)}, W^{(0)}, W^{(1)}$ are the set of simultaneous linear equations

$$
\begin{gather*}
\sum_{p}\left\{h_{p 0} k_{i p}-h_{i p} k_{p 0}\right\}+g_{i 0}=0, \quad i=1,2, \ldots, n \\
\sum_{p}\left\{h_{p 0}\left(k_{i j p}+k_{i p j}\right)+h_{p j} k_{i p}-j_{i p} k_{p j}-\left(h_{i p j}+h_{i j p}\right) k_{p 0}\right\}+g_{i j}=0 \\
i, j=1,2, \ldots, n \\
\sum_{p}\left\{h_{p j}\left(k_{i p k}+k_{i k p}\right)+h_{p k}\left(k_{i p j}+k_{i j p}\right)-h_{i p}\left(k_{p j k}+k_{p k j}\right)+\left(h_{p j k}+h_{p k j}\right) k_{i p}\right. \\
\left.-\left(h_{i p k}+h_{i k p}\right) k_{p j}-\left(h_{i p j}+h_{i j p}\right) k_{p k}\right\}+g_{i j k}+g_{i k j}=0 \\
i, j, k=1,2, \ldots, n \tag{2.10}
\end{gather*}
$$

In this "quadratic" approximation, each component of $g$ is uniquely determined by a single equation if one chooses $r$ to be a one-term homogeneous polynomial. Since the general solution of the equation is an arbitrary linear combination of these special solutions, one may make this choice without any loss of generality. In this linear combination, the coefficients may be arbitrary functions of the $k_{i m}$. We shall say that the generators $U_{m}$ in a collection are "independent" if no linear combination of them

$$
\sum c_{m} U_{m}
$$

is identically equal to zero when the coefficients $c_{m}$ in the linear combination are not functions of $\boldsymbol{x}$.

The remaining sections of this paper will make use of the quadratic approximation to the generators and the approximation to (2.8) obtained by dropping all $W^{(p)}$ with $p$ greater than 1 . We shall term this twofold approximation the "quadratric approximation". In paper II, we will investigate more accurate approximations to the generators and show that the quadratic approximation is of great utility.

In the two-species case, we obtain twelve equations $w_{i m}=0$ from the quadratic approximation to (2.8). Their general solution is a linear combination of twelve independent special solutions. Each special solution fixes a generator $U$, listed in table 2.1. The generators whose $h$ 's are of zero or first order in $x$ are exact solutions of (2.3).

Inspecting table 2.1, the reader will note that we have chosen the $U_{i m}$ to be of the form (here, $g^{i 0}$ is the $g$ vector of $U_{i 0}$, etc.)

$$
\begin{align*}
& U_{i 0}=\partial / \partial x_{i}+g^{i 0} \cdot \nabla_{k}, \quad U_{i j}=x_{j} \partial / \partial x_{i}+g^{i j} \cdot \nabla_{k} \\
& U_{i i j^{\prime}}=x_{j} x_{j^{\prime}} \partial / \partial x_{i}+g^{i j j^{\prime}} \cdot \nabla_{k} . \tag{2.11}
\end{align*}
$$

That is, eqs. (2.10) allow one to choose the action of each $U$ upon the species concentrations and then determine the action on the kinetic coefficients that is required to leave the kinetic equations invariant up through terms quadratic in the concentrations.

This procedure generalizes to systems of three or more species. As a result, one can easily obtain analogously exact and quadratically approximated invariance generators $U$ for kinetic systems (2.1) involving an arbitrary number of species. In the general case, the generators obtained with the aid of eqs. (2.10) are:

$$
\begin{aligned}
\dot{U}_{i 0}= & \partial / \partial x_{i}-\sum_{j} k_{i j} \partial / \partial k_{j 0}-\sum_{m \neq i} k_{j m i} \partial / \partial k_{j m}-2 \sum_{j} k_{j i i} \partial / \partial k_{j i} \\
U_{i i}= & x_{1} \partial / \partial x_{i}+k_{i 0} \partial / \partial k_{i 0}+\sum_{j \neq i} k_{i j} \partial / \partial k_{i j}-k_{i i i} \partial / \partial k_{i i i} \\
& +\sum_{j, m \neq i} k_{i j m} \partial / \partial k_{i j m}-\sum_{j \neq i} k_{j i} \partial / \partial k_{j i} \\
& -2 \sum_{j \neq i} k_{j i i} \partial / \partial k_{j i i}-\sum_{m, j \neq i} k_{j i m} \partial / \partial k_{j i m}
\end{aligned}
$$

For $j \neq i$ :
Table 2.1
Generators of invariance transformations

| Generator | $h$ |  |  |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $h_{1}$ | $h_{2}$ | $g_{10}$ | $g_{11}$ | $g_{12}$ | $g_{111}$ | $g_{112}$ | $g_{122}$ | $g_{20}$ | $g_{21}$ | $g_{22}$ | $g_{211}$ | $g_{212}$ |
| $U_{10}$ | 1 | 0 | $-k_{11}$ | $-2 k_{111}$ | $-k_{112}$ | 0 | 0 | 0 | $-k_{21}$ | $-2 k_{211}$ | $-k_{212}$ | 0 | $g_{222}$ |
| $U_{11}$ | $x_{1}$ | 0 | $k_{10}$ | 0 | $k_{12}$ | $-k_{111}$ | 0 | $k_{122}$ | 0 | $-k_{21}$ | 0 | $-2 k_{211}$ | $-k_{212}$ |
| $U_{12}$ | $x_{2}$ | 0 | $k_{20}$ | $k_{21}$ | $\left(k_{22}-k_{11}\right)$ | $k_{211}$ | $\left(k_{212}-2 k_{111}\right)$ | $\left(k_{222}-k_{112}\right)$ | 0 | 0 | $-k_{21}$ | 0 | 0 |
| $U_{11}$ | $x_{1}^{2}$ | 0 | 0 | $2 k_{10}$ | 0 | $k_{11}$ | $2 k_{12}$ | 0 | 0 | 0 | 0 | $-k_{21}$ | 0 |
| $U_{112}$ | $x_{1} x_{2}$ | 0 | 0 | $k_{20}$ | $k_{10}$ | $k_{21}$ | $k_{22}$ | $k_{12}$ | 0 | 0 | 0 | 0 | $-2 k_{211}$ |
| $U_{122}$ | $x_{2}^{2}$ | 0 | 0 | 0 | $2 k_{20}$ | 0 | $2 k_{21}$ | $\left(2 k_{22}-k_{11}\right)$ | 0 | 0 | 0 | 0 | 0 |
| $U_{20}$ | 0 | 1 | $-k_{12}$ | $-k_{112}$ | $-2 k_{122}$ | 0 | 0 | 0 | $-k_{22}$ | $-k_{212}$ | $-2 k_{222}$ | 0 | 0 |
| $U_{21}$ | 0 | $x_{1}$ | 0 | $-k_{12}$ | 0 | $-k_{112}$ | $-2 k_{122}$ | 0 | $k_{10}$ | $\left(k_{11}-k_{22}\right)$ | $k_{12}$ | $\left(k_{111}-k_{221}\right)$ | $\left(k_{112}-2 k_{222}\right)$ |
| $U_{22}$ | 0 | $x_{2}$ | 0 | 0 | $-k_{12}$ | 0 | $-k_{112}$ | $-2 k_{122}$ | $k_{20}$ | $k_{21}$ | 0 | $k_{122}$ |  |
| $U_{211}$ | 0 | $x_{1}^{2}$ | 0 | 0 | 0 | $-k_{12}$ | 0 | 0 | 0 | $2 k_{10}$ | 0 | $\left(2 k_{11}-k_{22}\right)$ | 0 |
| $U_{212}$ | 0 | $x_{1} x_{2}$ | 0 | 0 | 0 | 0 | $-k_{12}$ | 0 | 0 | $k_{20}$ | $k_{10}$ | $k_{21}$ | 0 |
| $U_{222}$ | 0 | $x_{2}^{2}$ | 0 | 0 | 0 | 0 | 0 | $-k_{12}$ | 0 | 0 | $2 k_{12}$ | $-k_{222}$ | 0 |

$$
\begin{align*}
U_{i j} & =x_{j} \partial / \partial x_{i}+k_{j 0} \partial / \partial k_{i 0}+k_{j i} \partial / \partial k_{i i}+\left(k_{i j}-k_{i i}\right) \partial / \partial k_{i j} \\
& +\sum_{m \neq i, j} k_{j m} \partial / \partial k_{i m}-\sum_{m \neq i} k_{m i} \partial / \partial k_{m j}+k_{j i i} \partial / \partial k_{i i i} \\
& +\left(k_{i j i}-2 k_{i i j}\right) \partial / \partial k_{i i j}+\left(k_{i j j}-k_{i i j}\right) \partial / \partial k_{i j j} \\
& +\sum_{m \neq i, j} k_{j i m} \partial / \partial k_{i i m}+\sum_{m \neq i, j}\left(k_{j j m}-k_{i i m}\right) \partial / \partial k_{i j m} \\
& -2 \sum_{m \neq i} k_{m i i} \partial / \partial k_{m i j}-\sum_{m, n \neq i} k_{m i n} \partial / \partial k_{m j n} \\
U_{i i i} & =x_{1} x_{i} \partial / \partial x_{i}+2 k_{i 0} \partial / \partial k_{i i}+k_{i i} \partial / \partial k_{i i i} \\
& +2 \sum_{j \neq i} k_{i j} \partial / \partial k_{i i j}-\sum_{j \neq i} k_{j i} \partial / \partial k_{j i i} \tag{2.12}
\end{align*}
$$

For $j \neq i$ :

$$
\begin{align*}
U_{i i j} & =x_{i} x_{j} \partial / \partial x_{i}+k_{j 0} \partial / \partial k_{i i}+k_{i 0} \partial / \partial k_{i j}+\sum k_{j m} \partial / \partial k_{i i m}+2 k_{i j} \partial / \partial k_{i j j} \\
& +\sum_{m \neq i, j} k_{i m} \partial / \partial k_{i j m}-\sum_{m \neq i} k_{m i} \partial / \partial k_{m i j} \\
U_{i j j} & =x_{j} x_{j} \partial / \partial x_{i}+2 k_{j 0} \partial / \partial k_{i j}+2 k_{j i} \partial / \partial k_{i i j}+\left(2 k_{j j}-k_{i i}\right) \partial / \partial k_{i j j} \\
& +2 \sum_{m \neq i, j} k_{j m} \partial / \partial k_{i j m}-\sum_{m \neq i} k_{m i} \partial / \partial k_{m j j} \tag{2.13}
\end{align*}
$$

For $i, j, j^{\prime}$ all different:

$$
\begin{align*}
U_{i j j^{\prime}} & =x_{j} x_{j^{\prime}} \partial / \partial x_{i}+k_{i 0} \partial / \partial k_{i j^{\prime}}+k_{j^{\prime} 0} \partial / \partial k_{i j}+k_{i j^{\prime}} \partial / \partial k_{i j^{\prime} j^{\prime}} \\
& +k_{j^{\prime} j} \partial / \partial k_{i j j}+\sum_{m \neq j^{\prime}} k_{j m} \partial / \partial k_{i m j^{\prime}} \\
& +\sum_{m \neq j} k_{j^{\prime} m} \partial / \partial k_{i j m}-\sum_{m} k_{m i} \partial / \partial k_{m i j^{\prime}} \tag{2.14}
\end{align*}
$$

In this list, the generators $U_{i 0}, U_{i i}$, and $U_{i j}$ exactly satisfy the determining eqs. (2.8). The generators $U_{i i i}, U_{i i j^{\prime}}$, and $U_{i j j^{\prime}}$ satisfy (2.8) in quadratic approximation.

## 3. Finite transformations

As mentioned earlier, corresponding to each generator $U$ there is an operator $\exp (a U)$ of finite transformations. One way to determine the effect of this upon each variable $x_{i}, k_{\mu}$ is to expand the exponential in powers of $a U$, carry out the indicated actions and sum the resulting series, which sometimes terminates, has evident recursiveness, or is recognizable as the MacLaurin expansion of a simple function. Often, a more practical method is to integrate the set of equations [4]:

$$
\begin{equation*}
\delta a=\frac{\delta x_{1}}{h_{1}}=\frac{\delta x_{2}}{h_{2}} \ldots=\frac{\delta k_{10}}{g_{10}}=\frac{\delta k_{20}}{g_{20}}=\frac{\delta k_{11}}{g_{11}} \ldots=\frac{\delta k_{2 i j^{\prime}}}{g_{2 j j^{\prime}}} \ldots \tag{3.1}
\end{equation*}
$$

When the $h$ are of the form we have chosen, the necessary integrations can all be carried out analytically.

Note that the only concentration altered by $T_{i m}(a)=\exp \left(a U_{i m}\right)$ is $x_{i}$. One finds using (3.1):

$$
\begin{array}{ll}
T_{i 0}(a) x_{i}=x_{i}+a, & T_{i i}(a) x_{i}=x_{i} e^{a} \\
T_{i j}(a) x_{i}=x_{i}+a x_{j}, & j \neq i, \\
T_{i i i}(a) x_{i}=x_{i} /\left(1-a x_{i}\right)  \tag{3.2}\\
T_{i i j}(a) x_{i}=x_{i} e^{a x_{j}}, & j \neq i,
\end{array} T_{i j j^{\prime}}(a) x_{i}=x_{i}+a x_{j} x_{j^{\prime}}, \quad i \neq j, j^{\prime} .
$$

The effect of each of the finite transformation operators on the kinetic parameters $k_{i m}$ are listed in table 2.2. As an example, one finds from table 2.2 that $T_{10}$ acting on $k_{10}$ gives $\bar{k}_{10}=k_{10}-a k_{11}$.

Because $T_{10}(a)$ and $T_{20}(b)$ leave $\dot{x}_{1}$ and $\dot{x}_{2}$ invariant, their action on the kinetic equations can be determined by replacing $x_{1}$ by $x_{1}+a$, or $x_{2}$ by $x_{2}+b$, in $r$ and determining the coefficients $c_{i j j^{\prime}}$ of the various powers $x_{j} x_{j^{\prime}}$ of the concentrations in the equation for $\dot{x}_{i}$. Then one finds $\bar{k}_{i j j^{\prime}}=c_{i j j^{\prime}}$. Because $T_{10}$ and $T_{20}$ carry out translations of $x$ while leaving the kinetic equations invariant in the generalized sense that the quadratic polynomic form of $r$ is preserved, we shall term them "invariant translation" operators.

In all the generators other than the $U_{i 0}$, the operator $\partial / \partial x_{i}$ is premultiplied by either $x_{i}$ or $x_{j}$. As a consequence, these generators vanish at the origin of $\boldsymbol{x}$. Because of this, the corresponding operators of finite transformations $T$ cannot move a point at the origin. If one lets $U$ be a linear combination of the generators in table 2.1, the finite transformations may be obtained by solving eqs. (3.1) de novo.

Before concluding this paper, we would call attention to some geometrical properties of our transformations. First note that the evolution generator $V$ is a special type of $U$ with $g=0$, and that the corresponding operator of finite trans-
Table 2.2(a)
Finite transformations of $\boldsymbol{k}$

|  | $\bar{k}_{10}$ | $\bar{k}_{11}$ | $\bar{k}_{12}$ | $\bar{k}_{111}$ | $\bar{k}_{112}$ | $\bar{k}_{122}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $T_{10}$ | $\begin{aligned} & k_{10}-a k_{11} \\ & +a^{2} k_{111} \end{aligned}$ | $k_{11}-2 a k_{111}$ | $k_{12}-a k_{112}$ | $k_{111}$ | $k_{112}$ | $k_{122}$ |
| $T_{12}$ | $e^{a_{k_{10}}}$ | $k_{11}$ | $e^{a_{k_{12}}}$ | $e^{-a_{k_{111}}}$ | $k_{112}$ | $e^{a} k_{122}$ |
| $T_{12}$ | $k_{10}+a k_{20}$ | $k_{11}+a k_{21}$ | $\begin{aligned} & k_{12}-a^{2} k_{21} \\ & +a\left(k_{22}-k_{11}\right) \end{aligned}$ | $k_{111}+a k_{211}$ | $\begin{aligned} & k_{112}-2 a^{2} k_{211} \\ & +a\left(k_{212}-2 k_{111}\right) \end{aligned}$ | $\begin{aligned} & k_{122}+a\left(k_{222}-k_{112}\right) \\ & +a^{2}\left(k_{111}-k_{212}\right)+a^{3} k_{211} \end{aligned}$ |
| $T_{111}$ | $k_{10}$ | $k_{11}+2 a k_{10}$ | $k_{12}$ | $k_{111}+a k_{11}+a^{2} k_{10}$ | $k_{112}+2 a k_{12}$ | $k_{122}$ |
| $T_{112}$ | $k_{10}$ | $k_{11}+a k_{20}$ | $k_{12}+a k_{10}$ | $k_{111}+a k_{21}$ | $k_{112}+a k_{22}$ | $k_{122}+a k_{12}+1 / 2 a^{2} k_{10}$ |
| $T_{122}$ | $k_{10}$ | $k_{11}$ | $k_{12}+2 a k_{20}$ | $k_{111}$ | $k_{112}+2 a k_{21}$ | $k_{122}+a\left(2 k_{22}-k_{11}\right)$ |

[^0]Table 2.2(c)
Finite transformations of $k$

|  | $\bar{k}_{10}$ | $\bar{k}_{11}$ | $\bar{k}_{12}$ | $\bar{k}_{111}$ | $\bar{k}_{112}$ | $\bar{k}_{122}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $T_{20}$ | $\begin{aligned} & k_{10}-a k_{12} \\ & +a^{2} k_{122} \end{aligned}$ | $k_{11}-a k_{112}$ | $k_{12}-2 a k_{122}$ | $k_{111}$ | $k_{112}$ | $k_{122}$ |
| $T_{21}$ | $k_{10}$ | $k_{11}-a k_{12}$ | $k_{12}$ | $\begin{aligned} & k_{111}-a k_{112} \\ & +a^{2} k_{122} \end{aligned}$ | $k_{112}-2 a k_{122}$ | $k_{122}$ |
| $T_{22}$ | $k_{10}$ | $k_{11}$ | $e^{-a_{k_{12}}}$ | $k_{11}$ | $e^{-a_{k_{12}}}$ | $e^{-2 a^{\prime} k_{122}}$ |
| $T_{211}$ | $k_{10}$ | $k_{11}$ | $k_{12}$ | $k_{111}-a k_{12}$ | $k_{112}$ | $k_{122}$ |
| $T_{212}$ | $k_{10}$ | $k_{11}$ | $k_{12}$ | $k_{111}$ | $k_{112}-a k_{12}$ | $k_{122}$ |
| $T_{222}$ | $k_{10}$ | $k_{11}$ | $k_{12}$ | $k_{111}$ | $k_{112}$ | $k_{122}-a k_{12}$ |

Table 2.2(d)
Finite transformations of $k$

|  | $\bar{k}_{20}$ | $\bar{k}_{21}$ | $\bar{k}_{22}$ | $\bar{k}_{211}$ | $\bar{k}_{212}$ | $\bar{k}_{222}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $T_{20}$ | $\begin{aligned} & k_{20}-a k_{22} \\ & +a^{2} k_{222} \end{aligned}$ | $k_{21}-a k_{212}$ | $k_{22}-2 a k_{222}$ | $k_{211}$ | $k_{212}$ | $k_{222}$ |
| $T_{21}$ | $k_{20}+a k_{10}$ | $\begin{aligned} & k_{21}-a^{2} k_{12} \\ & +a\left(k_{11}-k_{22}\right) \end{aligned}$ | $k_{22}+a k_{12}$ | $\begin{aligned} & k_{211}+a\left(k_{111}-k_{212}\right) \\ & +a^{2}\left(k_{222}-k_{112}\right) \\ & +a^{3} k_{122} \end{aligned}$ | $\begin{aligned} & k_{212}+a\left(k_{112}-2 k_{222}\right) \\ & -2 a^{2} k_{122} \end{aligned}$ | $k_{222}+a k_{122}$ |
| $T_{22}$ | $e^{a_{k_{20}}}$ | $e^{a_{k_{21}}}$ | $k_{22}$ | $e^{a} k_{211}$ | $k_{212}$ | $e^{-a_{k_{222}}}$ |
| $\mathrm{T}_{211}$ | $k_{20}$ | $k_{21}+2 a k_{10}$ | $k_{22}$ | $k_{211}+a\left(2 k_{11}-k_{22}\right)$ | $k_{212}+2 a k_{12}$ | $k_{222}$ |
| $T_{212}$ | $k_{20}$ | $k_{21}+a k_{20}$ | $k_{22}+a k_{10}$ | $\begin{aligned} & k_{211}+a k_{21} \\ & +1 / 2 a^{2} k_{20} \end{aligned}$ | $k_{212}+a k_{11}$ | $k_{222}+a k_{12}$ |
| $T_{222}$ | $k_{20}$ | $k_{21}$ | $k_{22}+2 a k_{20}$ | $k_{211}$ | $k_{212}+2 a k_{21}$ | $k_{222}+a k_{22}+a^{2} k_{20}$ |

formations $\exp (a V)$ becomes the time evolution operator if $a$ is replaced by $t$. Equations (3.1) then simply restate the kinetic equations (2.1). (Of course, $V$ is supposed known, while in the analysis just completed we have determined the $U$ 's allowed for a given $V$.) Now the operator $\exp (t V)$ evolves an initial point into a trajectory in the space of $x, k$ without changing the $k$ 's. Taken together, all these trajectories constitute a flow because the coefficients $r_{i}$ in (2.1) everywhere define a unique infinitesimal transformation $\exp (\delta t V)$. Each operator $\exp (a U)$ whose $U$ satisfies the determining eqs. (2.8) will take a point $P$ on such a trajectory and displace it in a transverse direction, by changing both $x$ and $k$, giving an image point $\bar{P}$. If, with the same value of $a, \exp (a U)$ acts on another point $P^{\prime}$ of the original trajectory, it will carry this


Fig. 2.1. Transformation flows $e^{a U_{x}}$ transverse to evolution flows $e^{t V_{x}}$. For each fixed value of the group parameter $a$, the transformation with generator $U$ carries the evolving concentrations $x_{i}(t)$ into an altered set of evolving concentrations. The transformed concentrations obey a set of elementary kinetic equations with altered rate constants.
into an image point $\bar{P}^{\prime}$. Because $\exp (\delta a U)$ everywhere defines a unique infinitesimal transformation and $U$ is not proportional to $V$, the collection of all these trajectories produced by $\exp (a U)$ constitutes a flow transverse to the flow produced by the evolution operator. As indicated in fig. 2.1, the evolution operator will evolve the image point $\bar{P}$ into a trajectory which will pass through $\bar{P}^{\prime}$ at the same time $t$ that $P$ is evolved into $P^{\prime}$. The proof of this observation follows from the fact that in deriving (2.8) we have required that $\partial U / \partial t$ vanishes. Thus, (2.4) becomes

$$
\begin{equation*}
[U, V]=0 \tag{3.3}
\end{equation*}
$$

which implies that

$$
\begin{equation*}
\exp (t V) \exp (a U)(\boldsymbol{x}, \boldsymbol{k})=\exp (a U) \exp (t V)(\boldsymbol{x}, \boldsymbol{k}) \tag{3.4}
\end{equation*}
$$

When the generator $U$ only approximately commutes with $V$, (3.4) will only hold approximately and the point obtained by transforming, then evolving, will not necessarily coincide with the point obtained by evolving, then transforming. This is the case for the generators $U_{i j j^{\prime}}$, for example.

## 4. Conclusions

Inspection of eqs. (2.8) shows that if $U^{(-1)}, U^{(0)}$, and $U^{(1)}$ all vanish, then $U$ does not act on the rate constants $k$. Thus, by determining all $U$ with nonvanishing $U^{(-1)}, U^{(0)}, U^{(1)}$ whose $T(a)$ transform elementary rate equations into elementary rate equations, we have found all $U$ generating one-parameter groups $\exp (a U)$ that transform elementary rate equations into elementary rate equations with different rate constants. The $U_{i}$ and the $U_{i j}$ have been determined exactly. In the $U_{i j j^{\prime}}$, the functions governing the transformation of species concentrations have been determined to second order in the concentrations, and the functions governing the transformation of the rate constants have been exactly determined.

Throughout this and the following paper, two one-parameter groups are composed by allowing the second to act on the result obtained from the action of the first. Thus, if

$$
\begin{equation*}
x_{1}^{\prime}=\exp \left(b U_{112}\right) x_{1}=x_{1} \exp \left(b x_{2}\right) \tag{4.1}
\end{equation*}
$$

and

$$
\begin{equation*}
x_{2}^{\prime}=\exp \left(a U_{222}\right) x_{2}=x_{2} /\left(1-a x_{2}\right) \tag{4.2}
\end{equation*}
$$

then the effect of the transformation $\exp \left(b U_{122}\right) \exp \left(a U_{222}\right)$ is to first shift the point with coordinates $x_{1}, x_{2}$ to the point with coordinates $\left(x_{1}, x_{2}^{\prime}\right)$. It then moves this to the point with coordinates $\left(x_{1}^{\prime}=x_{1} \exp \left(b x_{2}^{\prime}\right), x_{2}^{\prime}\right)$. Written as functions of the coordinates of the initial point, the coordinates of the final point are therefore

$$
\begin{equation*}
\left(x_{1} \exp \left(b\left\{x_{2} /\left(1-a x_{2}\right)\right\}\right), x_{2} /\left(1-a x_{2}\right)\right) \tag{4.3}
\end{equation*}
$$

From the one-parameter groups with operators $T_{\alpha}\left(a_{\alpha}\right)=\exp \left(a_{\alpha} U_{\alpha}\right)$, one may construct many-parameter groups $T_{\alpha \beta} \ldots\left(a_{\alpha}, a_{\beta}, \ldots\right)=\exp \left(a_{\alpha} U_{\alpha}\right) \exp \left(a_{\beta} U_{\beta}\right) \ldots$ whenever

$$
\begin{equation*}
\left[U_{\alpha}, U_{\beta}\right]=\sum c_{\alpha \beta}^{\nu} U_{\nu} \tag{4.4}
\end{equation*}
$$

for all $\alpha, \beta, \nu$. As all many-parameter groups may be obtained from one-parameter groups in this way, it may be concluded that our determination of the generators of all one-parameter groups that transform elementary rate equations into different elementary rate equations at once determines, exactly or approximately, all generators of many-parameter groups with this property. (In the following paper II, a list of such many-parameter groups is given for systems involving two chemical species.)

To conclude: In this paper, all generators of all one-parameter and all manyparameter groups of flows that transform elementary rate equations into elementary rate equations with different rate constants have been determined either exactly or approximately. A particularly simple generator basis has been chosen and the finite transformations obtained by exponentiating each generator have been determined.

## Acknowledgements

The authors wish to thank Guang-Hui Xu and Gordon Ballentine for assistance with the calculations. This research was supported by the Air Force Office of Scientific Research.

## References

[1] H. Rabitz, Chem. Rev. 87(1987)101.
[2] R. Cukoer, H. Levine and K. Schuler, J. Comp. Phys. 26(1978)1.
[3] C. Box, W. Hunter and J. Hunter, Statistics for Experimenters (Wiley, New York, 1978).
[4] C. Wulfman and H. Rabitz, J. Phys. Chem. 90(1986)2264.
[5] L.M. Hubbard, C. Wulfman and H. Rabitz, J. Phys. Chem. 90(1986)2273.
[6] Cf., for example, A. Cohen, An Introduction to the Lie Theory of One-Parameter Groups (Heath, Boston, 1911).


[^0]:    Table 2.2(b)
    Finite transformations of $k$

    |  | $\bar{k}_{20}$ | $\bar{k}_{21}$ | $\bar{k}_{22}$ | $\bar{k}_{211}$ | $\bar{k}_{212}$ | $\bar{k}_{222}$ |
    | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
    | $T_{10}$ | $k_{20}-a k_{21}$ | $k_{21}-2 a k_{211}$ | $k_{22}-a k_{212}$ | $k_{211}$ | $k_{212}$ | $k_{222}$ |
    |  | $+a^{2} k_{211}$ |  |  |  |  |  |
    | $T_{11}$ | $k_{20}$ | $e^{-a} k_{21}$ | $e_{22}$ | $e^{-2 a k_{211}}$ | $e^{-2 a k_{212}}$ | $k_{222}$ |
    | $T_{12}$ | $k_{20}$ | $k_{21}$ | $k_{22}-a k_{21}$ | $k_{211}$ | $k_{212}-2 a k_{211}$ | $k_{222}-a k_{212}+a^{2} k_{211}$ |
    | $T_{111}$ | $k_{20}$ | $k_{21}$ | $k_{22}$ | $k_{211}-a k_{21}$ | $k_{212}$ | $k_{222}$ |
    | $T_{162}$ | $k_{20}$ | $k_{21}$ | $k_{22}$ | $k_{211}$ | $k_{212}-a k_{21}$ | $k_{222}$ |
    | $T_{122}$ | $k_{20}$ | $k_{21}$ | $k_{22}$ | $k_{211}$ | $k_{222}-a k_{21}$ |  |

